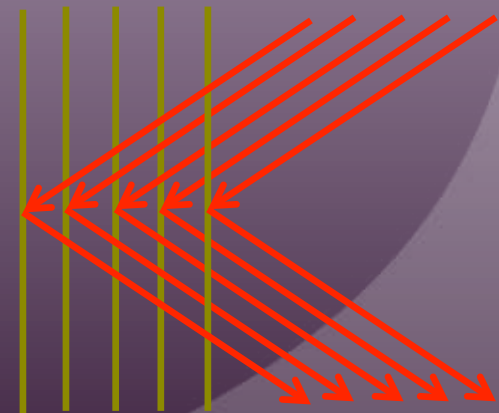


# Electron Diffraction with Crystals

# Bragg and Thompson

- Bragg was using X-rays, and looking at the interference between crystal planes at different depths.
- Thompson was using high energy electrons and a polycrystalline foil was also seeing contributions from many crystal planes (with many orientations).
- cf. E&R pg 59



# Davisson-Germer Experiment

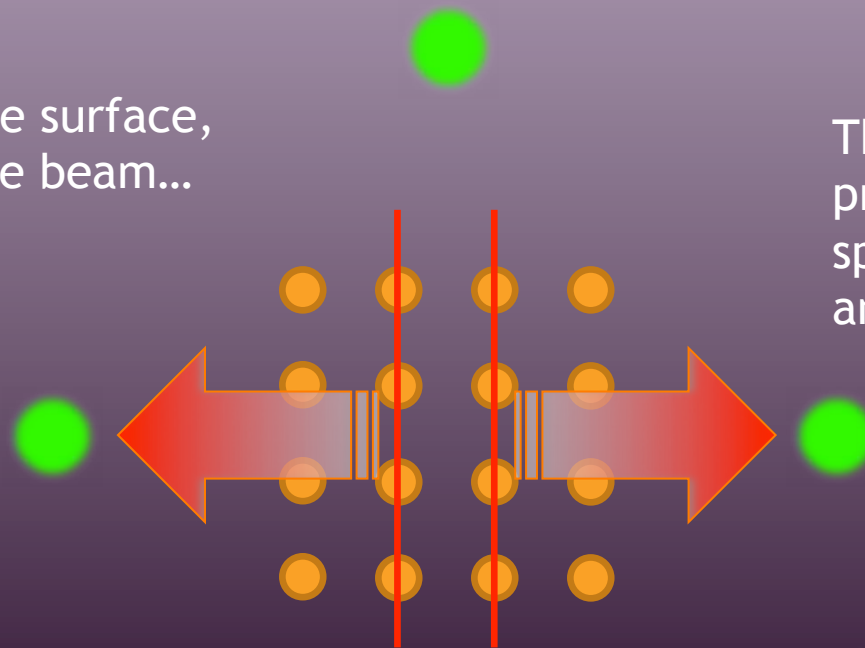
- Used low-energy electrons (54 eV)
- This technique, later developed into Low-Energy Electron Diffraction ([http://en.wikipedia.org/wiki/Low-energy\\_electron\\_diffraction](http://en.wikipedia.org/wiki/Low-energy_electron_diffraction)), only samples the surface layer of the target material
- So, why the diagonal spots?
- And why there?



# Horizontal and Vertical Spots

- Separation between vertical and horizontal planes is  $D$
- Angle of spot (due to constructive interference) is  $\theta$

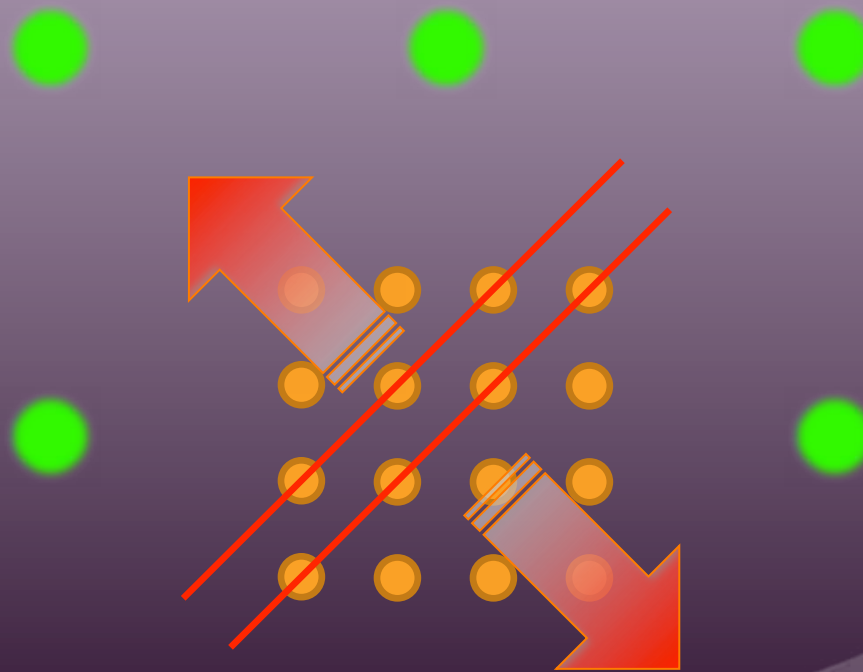
Looking at the surface,  
as seen by the beam...



The vertical planes  
produce the bright  
spots to the left  
and right.

# Diagonal Spots

- Separation between diagonal planes is  $D/\sqrt{2}$
- The planes are closer together than the vertical and horizontal planes, so  $\theta$  is larger by  $\sqrt{2}$



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8.04 Quantum Physics I  
Spring 2013

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