

# **MEDICAL GEOLOGY/GEOCHEMISTRY: An Exposure**

**PILLALAMARRI ILA**

**Earth Atmospheric & Planetary Sciences  
Neutron Activation Analysis Laboratory  
Massachusetts Institute of Technology  
Cambridge, MA 02139**

**IAP 2006: 12.091 Credit Course: January 9 - 25, 2006**

**Session 1, January 9, 2006**

# Detailed course work

The course work involves the following:

1. January 9, 11, 16, 18, 23 10 AM to 12 PM  
5 sessions each of 2 hours  
25%
2. Review Quiz– 5 Questions  
20%
3. Project  
Literature Survey – Writing a report  
30%
4. Project Presentation  
25%

Required percentage to pass this course is 85%  
Grading: P/F

Session 1

January 9, 2006

## Objective

### Introduction:

Definitions and terminology of Medical  
Geology/Geochemistry

### Classification of elements:

Periodic table of elements

Major, minor and trace Elements: Geological and  
biological materials

Geochemical classification

### Elemental link between geosphere and biosphere:

An attempt to understanding

Essential and non-essential elements with  
reference to human health

### Selection of Elements for Report and Presentation

# **Introduction**

**Definitions & Terminology  
Of  
Medical  
Geology/Geochemistry**

# Definitions – Terminology



## **Geology** definition

- Scientific study of the origin, history, and structure of the earth.
- Structure of a specific region of the earth's crust.
- Scientific study of the origin, history, and structure of the solid matter of a celestial body.

## What is **geology**?

“Geology, the study of the earth, employs the methods of other sciences, as well as those unique to geology, to investigate the nature, processes, and history of the earth.”

<http://www.udel.edu/geology/define.html>

## Definitions – Terminology ...



## Geochemistry

- The scientific study of the composition and alterations of the solid matter of the earth or a celestial body.
- Scientific study of chemical processes and reactions forming the rocks and soils, and the cyclic processes that transport the Earth's chemical components in time and space.

Some important fields of geochemistry are:

- **Isotope geochemistry**: Scientific determination of the relative and absolute concentrations of the elements and their isotopes inside the earth and on the surface, examining the distribution and movements of elements in different parts of the earth such as crust, mantle etc., and also determining the distribution and movement of minerals.
- **Cosmogeochimistry**: Chemistry of the composition of elements and their isotopes in the **cosmos**.
- **Organic geochemistry**: Scientific study of processes and compounds that are derived from living or once-living organisms.

# Definitions – Terminology ...



## **Environmental Geochemistry**

is the study of the geologic processes that interact strongly with the materials present in the Earth's crust.

### **The geologic processes:**

- weathering of rocks,
- soil formation,
- solutes in surface and ground water,
- global atmospheric transport of particulate matter,
- global cycling of contaminant products.

# Definitions – Terminology ...



**“geomedicine”** or  
**“geographic medicine”**

In 1930s , was the science using the geographical and cartographical methods to explain the results of medical research.

The term was **redefined** in 1990 as  
“the science dealing with the influence of ordinary environmental factors on the geographic distribution of health Problems in man and animals”.

**- J. Lag**

Reference: General Survey of Geomedicine in Geomedicine, pp 1-24, J. Lag CRC Press, Boca Raton, FL, 1990.



# Definitions – Terminology ...

**Medical Geology  
/Geochemistry**

**Medical geochemistry** looks at the effects of geochemical processes and geological factors on the health of humans and plants.

**Geomedicine  
/Geographic Medicine**

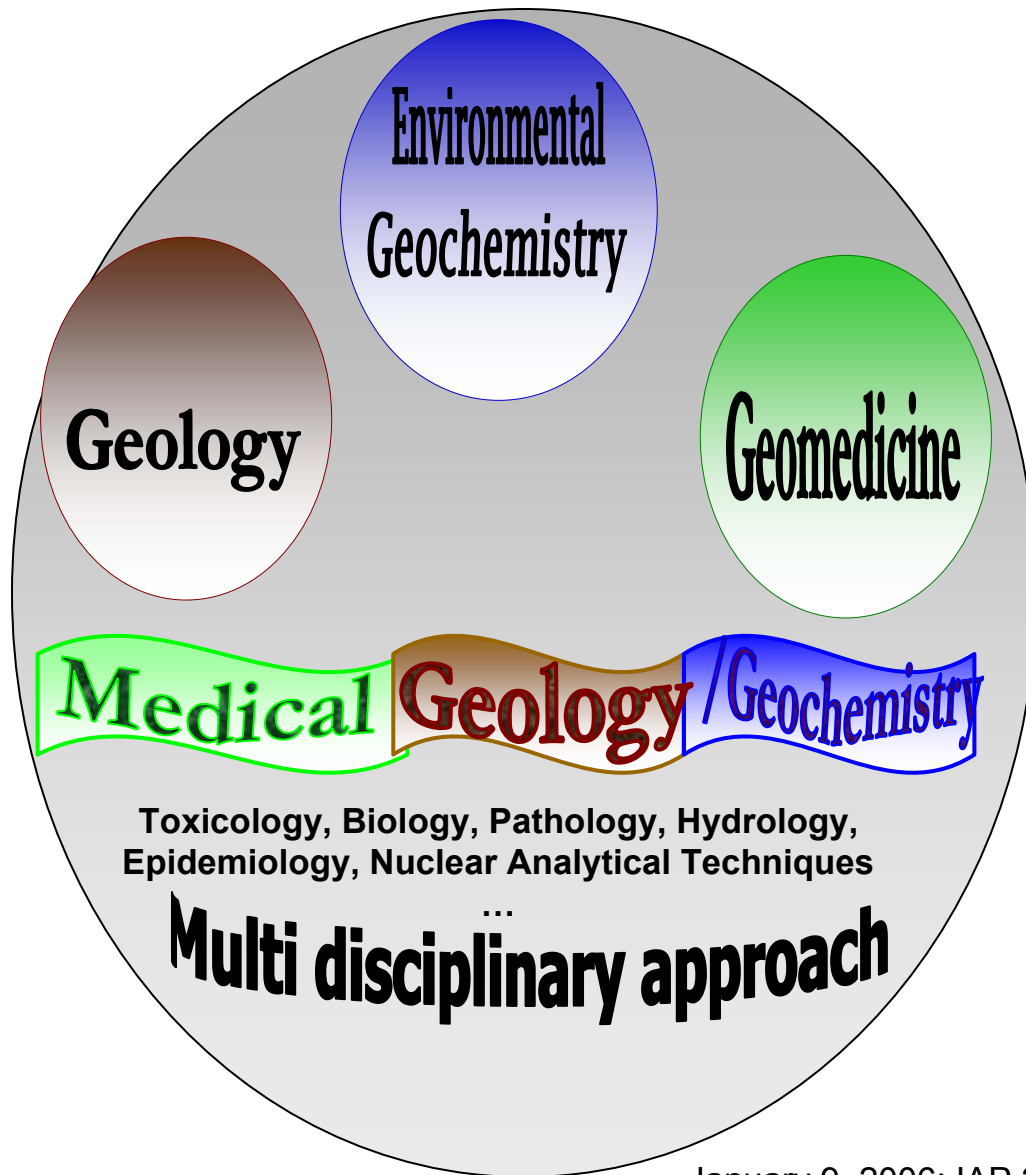
“**Medical geography** looks at the geographical distribution of disease while **not focusing** on the underlying geology. It examines the causal associations between specific diseases and the physical and social environments.”

References: Selinus 2002, Finkelman et al 2001 ; Lag 1990; Medical Geology: new relevance in the earth sciences. C. A. Bowman, P. T. Bobrowsky, O. Selinus Episodes Vol. 26(4) 270-278, 2003

**Medical geology/geochemistry and geographic medicine definitions**

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# Definitions – Terminology ...



## **Medical Geology/Geochemistry**

is the study of interaction between abundances of elements and isotopes and the health of humans and plants.

# Classification of Elements

- Periodic Table of Elements
- Major-Minor-Trace Elements
- Geochemical Classification
- Essential and Non-Essential

# Classification of elements

Before I proceed to explain the relationship between the concentration of elements and human nutrition, first I like to review the classification of elements.

Different studies classified the elements differently.

The well known Periodic Table of Elements shows the grouping of elements based on their chemical characteristics.

# Figure 1. Periodic Table of Elements

|   | 1                  | 2  | 3  | 4                   | 5                   | 6                  | 7                  | 8                  | 9                  | 10                 | 11                 | 12                  | 13                 | 14                  | 15                  | 16                  | 17                  | 18                  |
|---|--------------------|--|--|---------------------|---------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|---------------------|--------------------|---------------------|---------------------|---------------------|---------------------|---------------------|
| 1 | 1<br>H<br>1.008    | <div style="display: flex; justify-content: space-between;"> <div> <p>Alkali Metals</p> <p>Atomic Number ← 27</p> <p>Symbol ← Co</p> <p>Atomic Weight ← 58.933</p> </div> <div> <p>Noble Gases</p> </div> </div> |  |                     |                     |                    |                    |                    |                    |                    |                    |                     |                    |                     |                     |                     |                     | 2<br>He<br>4.003    |
| 2 | 3<br>Li<br>6.941   | 4<br>Be<br>9.012   | <div style="display: flex; justify-content: space-between;"> <div> <p>Alkaline Earth Metals</p> </div> <div> <p>Solids</p> </div> <div> <p>Non Metals</p> </div> </div>                          |                     |                     |                    |                    |                    |                    |                    |                    |                     | 5<br>B<br>10.811   | 6<br>C<br>12.011    | 7<br>N<br>14.007    | 8<br>O<br>15.999    | 9<br>F<br>18.998    | 10<br>Ne<br>20.180  |
| 3 | 11<br>Na<br>22.99  | 12<br>Mg<br>24.305   | <div style="display: flex; justify-content: space-between;"> <div> <p>Transition Metals</p> </div> <div> <p>Gases</p> </div> <div> <p>Liquids</p> </div> <div> <p>Other Metals</p> </div> </div> |                     |                     |                    |                    |                    |                    |                    |                    |                     | 13<br>Al<br>26.982 | 14<br>Si<br>28.086  | 15<br>P<br>30.974   | 16<br>S<br>32.060   | 17<br>Cl<br>35.453  | 18<br>Ar<br>39.948  |
| 4 | 19<br>K<br>39.098  | 20<br>Ca<br>40.08  | 21<br>Sc<br>44.956   | 22<br>Ti<br>47.88   | 23<br>V<br>50.94    | 24<br>Cr<br>51.996 | 25<br>Mn<br>54.938 | 26<br>Fe<br>55.847 | 27<br>Co<br>58.933 | 28<br>Ni<br>58.69  | 29<br>Cu<br>63.546 | 30<br>Zn<br>65.39   | 31<br>Ga<br>69.72  | 32<br>Ge<br>72.61   | 33<br>As<br>74.922  | 34<br>Se<br>78.96   | 35<br>Br<br>79.904  | 36<br>Kr<br>83.80   |
| 5 | 37<br>Rb<br>85.47  | 38<br>Sr<br>87.82  | 39<br>Y<br>88.906  | 40<br>Zr<br>91.22   | 41<br>Nb<br>92.906  | 42<br>Mo<br>95.94  | 43<br>Tc<br>(98)   | 44<br>Ru<br>101.07 | 45<br>Rh<br>102.91 | 46<br>Pd<br>106.4  | 47<br>Ag<br>107.87 | 48<br>Cd<br>112.41  | 49<br>In<br>114.82 | 50<br>Sn<br>118.71  | 51<br>Sb<br>121.75  | 52<br>Te<br>127.60  | 53<br>I<br>126.90   | 54<br>Xe<br>131.29  |
| 6 | 55<br>Cs<br>132.91 | 56<br>Ba<br>137.33   | 57<br>to<br>71   | 72<br>Hf<br>178.49  | 73<br>Ta<br>180.95  | 74<br>W<br>183.85  | 75<br>Re<br>186.21 | 76<br>Os<br>190.20 | 77<br>Ir<br>192.20 | 78<br>Pt<br>195.08 | 79<br>Au<br>196.97 | 80<br>Hg<br>200.59  | 81<br>Tl<br>204.88 | 82<br>Pb<br>207.20  | 83<br>Bi<br>208.98  | 84<br>Po<br>(209)   | 85<br>At<br>(210)   | 86<br>Rn<br>222.02  |
| 7 | 87<br>Fr<br>(223)  | 88<br>Ra<br>226.03   | 89<br>to<br>103  | 104<br>Rf<br>261.10 | 105<br>Db<br>262.11 | 106<br>Sg<br>(266) | 107<br>Bh<br>(264) | 108<br>Hs<br>(277) | 109<br>Mt<br>(268) | 110<br>Ds<br>(271) | 111<br>Rg<br>(272) | 112<br>Uub<br>(285) |                    | 114<br>Uuq<br>(289) |                     | 116<br>Uuh<br>(289) |                     |                     |
|   | 57<br>to<br>71     |  | Lanthanides  | 57<br>La<br>138.91  | 58<br>Ce<br>140.12  | 59<br>Pr<br>140.91 | 60<br>Nd<br>144.24 | 61<br>Pm<br>(145)  | 62<br>Sm<br>150.36 | 63<br>Eu<br>151.96 | 64<br>Gd<br>157.25 | 65<br>Tb<br>158.93  | 66<br>Dy<br>162.50 | 67<br>Ho<br>164.93  | 68<br>Er<br>167.26  | 69<br>Tm<br>168.93  | 70<br>Yb<br>173.04  | 71<br>Lu<br>174.97  |
|   | 89<br>to<br>103    |  | Actinides  | 89<br>Ac<br>227.03  | 90<br>Th<br>232.04  | 91<br>Pa<br>231.04 | 92<br>U<br>238.03  | 93<br>Np<br>237.05 | 94<br>Pu<br>244.06 | 95<br>Am<br>243.06 | 96<br>Cm<br>247.07 | 97<br>Bk<br>247.07  | 98<br>Cf<br>251.08 | 99<br>Es<br>252.08  | 100<br>Fm<br>257.10 | 101<br>Md<br>258.10 | 102<br>No<br>259.10 | 103<br>Lr<br>262.11 |

Note: The symbol for element 105 is Db [Ref. CRC Handbook of Chemistry and Physics, 86th Edition, 2005-2006].

The symbol Ha, also, is in use for element 105. [Ref. <http://periodic.lanl.gov>]

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# Figure 2. Periodic Table of Elements

Major – Minor and Trace elements classification with reference to biosphere

|   | 1                                | 2                                | 3                                | 4                                | 5                                 | 6                                 | 7                                | 8                                | 9                                | 10                               | 11                               | 12                               | 13                                | 14                                | 15                                | 16                                | 17                                | 18                                |
|---|----------------------------------|----------------------------------|----------------------------------|----------------------------------|-----------------------------------|-----------------------------------|----------------------------------|----------------------------------|----------------------------------|----------------------------------|----------------------------------|----------------------------------|-----------------------------------|-----------------------------------|-----------------------------------|-----------------------------------|-----------------------------------|-----------------------------------|
| 1 | <b>1</b><br><b>H</b><br>1.008    | Major Element                    |                                  |                                  |                                   |                                   |                                  |                                  |                                  |                                  |                                  |                                  |                                   |                                   |                                   |                                   | <b>2</b><br><b>He</b><br>4.003    |                                   |
| 2 | <b>3</b><br><b>Li</b><br>6.941   | <b>4</b><br><b>Be</b><br>9.012   |                                  |                                  |                                   |                                   |                                  |                                  |                                  |                                  |                                  |                                  | <b>5</b><br><b>B</b><br>10.811    | <b>6</b><br><b>C</b><br>12.011    | <b>7</b><br><b>N</b><br>14.007    | <b>8</b><br><b>O</b><br>15.999    | <b>9</b><br><b>F</b><br>18.998    | <b>10</b><br><b>Ne</b><br>20.180  |
| 3 | <b>11</b><br><b>Na</b><br>22.99  | <b>12</b><br><b>Mg</b><br>24.305 |                                  |                                  |                                   |                                   |                                  |                                  |                                  |                                  |                                  |                                  | <b>13</b><br><b>Al</b><br>26.982  | <b>14</b><br><b>Si</b><br>28.086  | <b>15</b><br><b>P</b><br>30.974   | <b>16</b><br><b>S</b><br>32.060   | <b>17</b><br><b>Cl</b><br>35.453  | <b>18</b><br><b>Ar</b><br>39.948  |
| 4 | <b>19</b><br><b>K</b><br>39.098  | <b>20</b><br><b>Ca</b><br>40.08  | <b>21</b><br><b>Sc</b><br>44.956 | <b>22</b><br><b>Ti</b><br>47.88  | <b>23</b><br><b>V</b><br>50.94    | <b>24</b><br><b>Cr</b><br>51.996  | <b>25</b><br><b>Mn</b><br>54.938 | <b>26</b><br><b>Fe</b><br>55.847 | <b>27</b><br><b>Co</b><br>58.933 | <b>28</b><br><b>Ni</b><br>58.69  | <b>29</b><br><b>Cu</b><br>63.546 | <b>30</b><br><b>Zn</b><br>65.39  | <b>31</b><br><b>Ga</b><br>69.72   | <b>32</b><br><b>Ge</b><br>72.61   | <b>33</b><br><b>As</b><br>74.922  | <b>34</b><br><b>Se</b><br>78.96   | <b>35</b><br><b>Br</b><br>79.904  | <b>36</b><br><b>Kr</b><br>83.80   |
| 5 | <b>37</b><br><b>Rb</b><br>85.47  | <b>38</b><br><b>Sr</b><br>87.82  | <b>39</b><br><b>Y</b><br>88.906  | <b>40</b><br><b>Zr</b><br>91.22  | <b>41</b><br><b>Nb</b><br>92.906  | <b>42</b><br><b>Mo</b><br>95.94   | <b>43</b><br><b>Tc</b><br>(98)   | <b>44</b><br><b>Ru</b><br>101.07 | <b>45</b><br><b>Rh</b><br>102.91 | <b>46</b><br><b>Pd</b><br>106.4  | <b>47</b><br><b>Ag</b><br>107.87 | <b>48</b><br><b>Cd</b><br>112.41 | <b>49</b><br><b>In</b><br>114.82  | <b>50</b><br><b>Sn</b><br>118.71  | <b>51</b><br><b>Sb</b><br>121.75  | <b>52</b><br><b>Te</b><br>127.60  | <b>53</b><br><b>I</b><br>126.90   | <b>54</b><br><b>Xe</b><br>131.29  |
| 6 | <b>55</b><br><b>Cs</b><br>132.91 | <b>56</b><br><b>Ba</b><br>137.33 | <b>57</b><br>to<br><b>71</b>     | <b>72</b><br><b>Hf</b><br>178.49 | <b>73</b><br><b>Ta</b><br>180.95  | <b>74</b><br><b>W</b><br>183.85   | <b>75</b><br><b>Re</b><br>186.21 | <b>76</b><br><b>Os</b><br>190.20 | <b>77</b><br><b>Ir</b><br>192.20 | <b>78</b><br><b>Pt</b><br>195.08 | <b>79</b><br><b>Au</b><br>196.97 | <b>80</b><br><b>Hg</b><br>200.59 | <b>81</b><br><b>Tl</b><br>204.88  | <b>82</b><br><b>Pb</b><br>207.20  | <b>83</b><br><b>Bi</b><br>208.98  | <b>84</b><br><b>Po</b><br>(209)   | <b>85</b><br><b>At</b><br>(210)   | <b>86</b><br><b>Rn</b><br>222.02  |
| 7 | <b>87</b><br><b>Fr</b><br>(223)  | <b>88</b><br><b>Ra</b><br>226.03 | <b>89</b><br>to<br><b>103</b>    | <b>89</b><br>to<br><b>103</b>    | <b>104</b><br><b>Rf</b><br>261.10 | <b>105</b><br><b>Db</b><br>262.11 | <b>106</b><br><b>Sg</b><br>(266) | <b>107</b><br><b>Bh</b><br>(264) | <b>108</b><br><b>Hs</b><br>(277) | <b>109</b><br><b>Mt</b><br>(268) | <b>110</b><br><b>Ds</b><br>(271) | <b>111</b><br><b>Rg</b><br>(272) | <b>112</b><br><b>Uub</b><br>(285) |                                   | <b>114</b><br><b>Uuq</b><br>(289) |                                   | <b>116</b><br><b>Uuh</b><br>(289) |                                   |
|   | Toxic Element                    |                                  |                                  |                                  |                                   |                                   |                                  |                                  |                                  |                                  |                                  |                                  |                                   |                                   |                                   |                                   |                                   |                                   |
|   | <b>57</b><br>to<br><b>71</b>     | Lanthanides                      |                                  |                                  |                                   |                                   |                                  |                                  |                                  |                                  |                                  |                                  |                                   |                                   |                                   |                                   |                                   | <b>71</b><br><b>Lu</b><br>174.97  |
|   | <b>89</b><br>to<br><b>103</b>    | Actinides                        |                                  |                                  |                                   |                                   |                                  |                                  |                                  |                                  |                                  |                                  |                                   |                                   |                                   |                                   |                                   | <b>103</b><br><b>Lr</b><br>262.11 |
|   |                                  | <b>57</b><br><b>La</b><br>138.91 | <b>58</b><br><b>Ce</b><br>140.12 | <b>59</b><br><b>Pr</b><br>140.91 | <b>60</b><br><b>Nd</b><br>144.24  | <b>61</b><br><b>Pm</b><br>(145)   | <b>62</b><br><b>Sm</b><br>150.36 | <b>63</b><br><b>Eu</b><br>151.96 | <b>64</b><br><b>Gd</b><br>157.25 | <b>65</b><br><b>Tb</b><br>158.93 | <b>66</b><br><b>Dy</b><br>162.50 | <b>67</b><br><b>Ho</b><br>164.93 | <b>68</b><br><b>Er</b><br>167.26  | <b>69</b><br><b>Tm</b><br>168.93  | <b>70</b><br><b>Yb</b><br>173.04  | <b>71</b><br><b>Lu</b><br>174.97  |                                   |                                   |
|   |                                  | <b>89</b><br><b>Ac</b><br>227.03 | <b>90</b><br><b>Th</b><br>232.04 | <b>91</b><br><b>Pa</b><br>231.04 | <b>92</b><br><b>U</b><br>238.03   | <b>93</b><br><b>Np</b><br>237.05  | <b>94</b><br><b>Pu</b><br>244.06 | <b>95</b><br><b>Am</b><br>243.06 | <b>96</b><br><b>Cm</b><br>247.07 | <b>97</b><br><b>Bk</b><br>247.07 | <b>98</b><br><b>Cf</b><br>251.08 | <b>99</b><br><b>Es</b><br>252.08 | <b>100</b><br><b>Fm</b><br>257.10 | <b>101</b><br><b>Md</b><br>258.10 | <b>102</b><br><b>No</b><br>259.10 | <b>103</b><br><b>Lr</b><br>262.11 |                                   |                                   |

Note: Based on Figure 1. pp XI, Essentials of Medical Geology. The symbol for element 105 is Db [Ref. CRC Handbook of Chemistry and Physics, 86th Edition, 2005-2006]. The symbol Ha, also, is in use for element 105 [Ref. <http://periodic.lanl.gov>].

# Classification of elements :

## Major – Minor – Trace

**Major, Minor and Trace element concentrations:**

- **Major: Concentrations exceeding 1% by mass;**  
**1% = 1g / 100g**
- **Minor: Concentrations in the range 0.1% to 1.0% by mass**
- **Trace : Concentrations less than 0.1% by mass**

**ppm     $\mu\text{g/g}$  (micro gram/gram)  $10^{-6}$  g/g**

**ppb     $\text{ng/g}$  (nano gram/gram)  $10^{-9}$  g/g**

**ppt     $\text{pg/g}$  (pico gram/gram)  $10^{-12}$  g/g**

**ppf     $\text{fg/g}$  (femto gram/gram)  $10^{-15}$  g/g**

**ppa     $\text{ag/g}$  (atto gram/ gram)  $10^{-18}$  g/g**

# Classification of elements

- Major – Minor – Trace and Toxic concepts are context dependent.
- Major – Minor – Trace abundances of elements are different for different category of materials.

For example they are different for geological and biological materials.



Table 1.  
Major-Minor abundances of different elements in  
two different categories of materials

| Element   | Earth's Crust | Human Body |
|-----------|---------------|------------|
|           | Abundance     | Abundance  |
| Oxygen    | 46.6%         | 65.4%      |
| Silicon   | 27.7%         | 60 ppm     |
| Iron      | 5.0%          | 260 ppm    |
| Calcium   | 3.6%          | 1.4%       |
| Sodium    | 2.8%          | 0.14%      |
| Potassium | 2.6%          | 0.34%      |
| Magnesium | 2.1%          | 0.5%       |

Based on Figure 1.3, pp 6, An Introduction to Environmental Chemistry; Table 1, pp 116, Essentials of Medical Geology.

# Classification of elements- Geochemical

Grouping the elements according to their geochemical associations:

-phile means 'forming' or 'loving'.

**Atmo**phile related to atmosphere.

**Chalco**phile - chalco means copper.

**Litho**phile – litho means stone – crustal.

**Sidero**phile – sidero means iron.

# Classification of elements Geochemical

**Table 2 . Geochemical Empirical Classification of Elements**

| <b>Classification</b> | <b>Brief Characteristics</b>  | <b>Main Elements</b>  |
|-----------------------|---|---|
| <b>Atmophile</b>      | Predominant in air  | H, He, Hg, N, O and other noble gases and C ( as CO <sub>2</sub> )  |
| <b>Chalcophile</b>    | 'Form sulfides, arsenides, selenides, tellurides;<br>Sources of ore minerals for nonferrous metals'   | Ag, As, Cd, Cu, Hg, Pb, S, Te, Zn   |
| <b>Lithophile</b>     | 'Form silicates, aluminosilicates, oxides, carbonates, sulfates, halides, phosphates and vandates among other mineral forms in the natural environment' | Al, Ba, Ca, Cs, Li, K, Mg, Na, Rb, Sr, REE<br>REE: La, Ce, Pr, Nd, Pm, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu |
| <b>Siderophile</b>    | 'Form alloys with iron and these are important sources of platinum group metals and gold'.  | Au, C, Co, Fe, Ge, Mo, Ni, P, Pt, Sn  |

Empirical means based on observation, valid information, not from theory

Based on Table 1, pp 26, Essentials of Medical Geology.

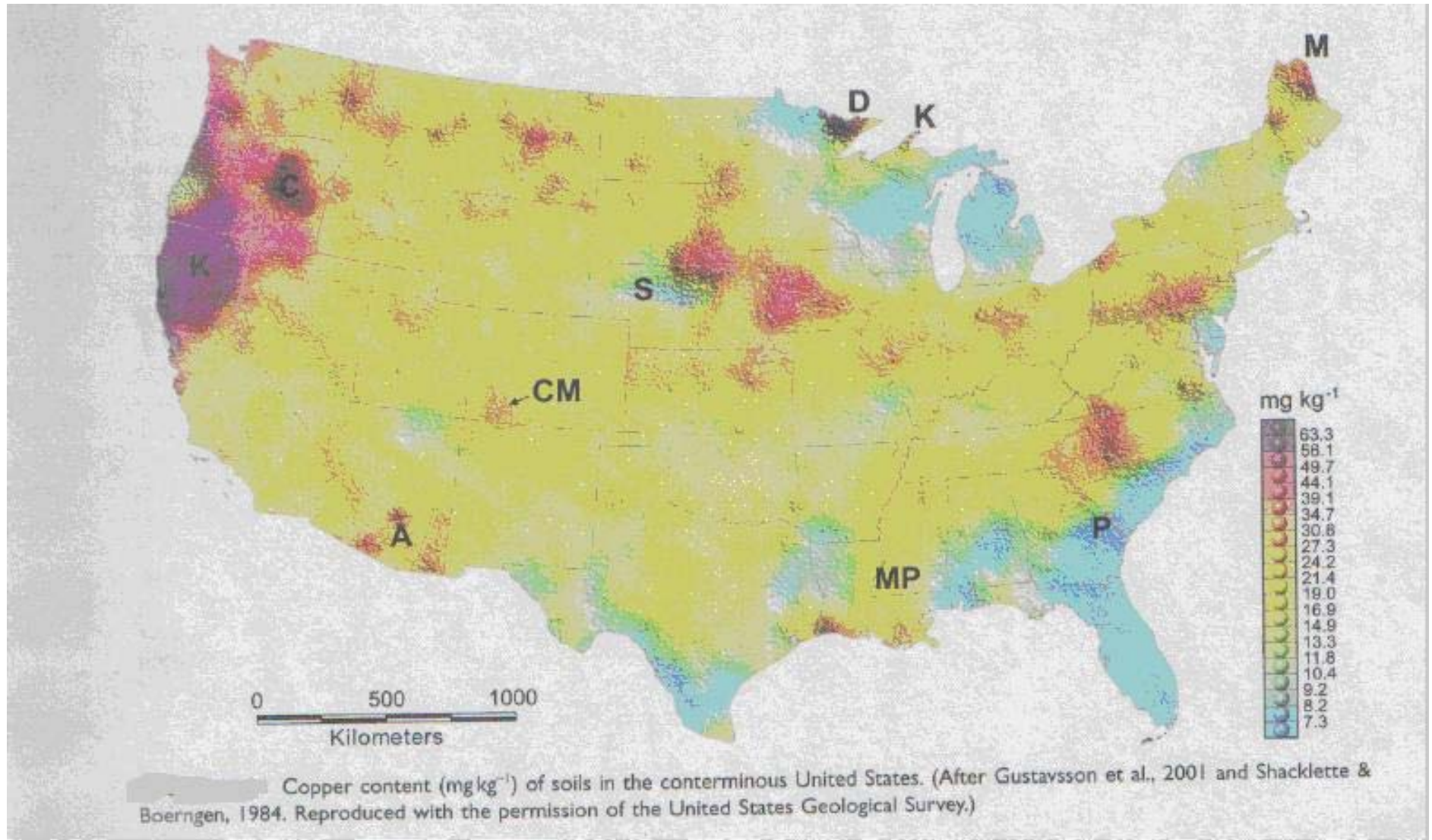
# What is the purpose of all these classifications?

The purpose is

- They provide the basis for the understanding of chemical diversity of the geological and biological materials.
- This understanding leads to further understanding of the interlink between the geosphere and biosphere and ultimately the influences on the human health.

## Chemical Variability of the Earth ...

Figure 3. Concentration variation of a single element in different regions.



Courtesy of USGS.

Reference: Geochemical land scapes of the conterminous United States – New map presentations for 22 elements, N. Gustavsson, B. Bolviken, D. B. Smith, and R. C. Sverson, U. S. Geol. Surv. Bull. 1645 (2001) 38.

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# Classification of elements : Major – Minor -Trace

## Chemical Variability of the Earth

Table 3. Concentration variation of a single element in different regions.

### Important aspects

- 1) High levels in the Northwest vs. Low levels in the Southeast
- 2) High levels in Minnesota – associated with Duluth Gabbro
- 3) High levels in Arizona where copper is mined
- 4) There is no one average background level.
- 5) Background values are for contiguous regional areas
- 6) Background levels are ranges reflecting the natural heterogeneity

| Variation of Cu concentration |            |
|-------------------------------|------------|
|                               | Cu (mg/kg) |
| Earth's Crust                 | 55-63      |
| Continental Crust             | 25-50      |
| Igneous                       |            |
| Ultramafic                    | 10         |
| Mafic                         | 87         |
| Sedimentary rocks             |            |
| Sandstone                     | 15         |
| Limestone                     | 4          |
| Shale                         | 45         |
| Black shale                   | 50-200     |

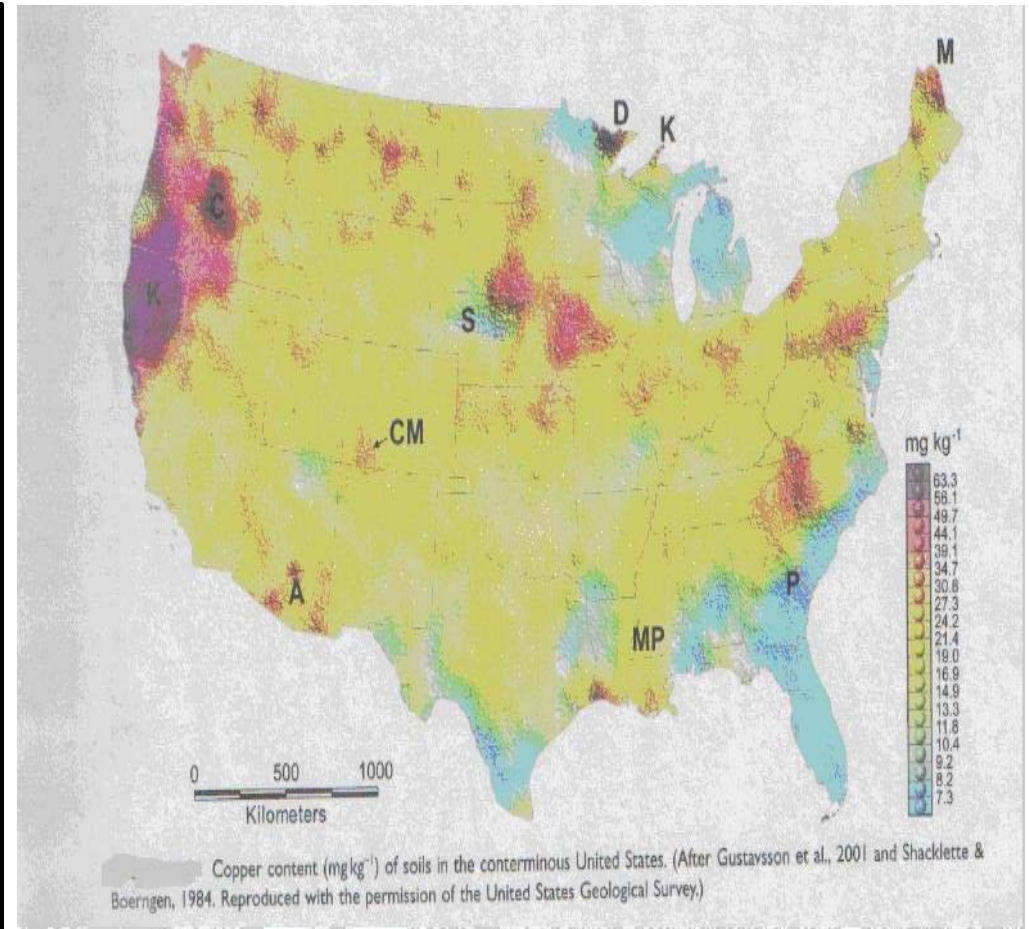


Figure 3. Concentration variation of a single element in different regions.

Courtesy of USGS.

Reference: Geochemical land scapes of the conterminous United States – New map presentations for 22 elements, N. Gustavsson, B. Bolviken, D. B. Smith, and R. C. Sverson, U. S. Geol. Surv. Bull. 1645 (2001) 38.

January 9, 2006: IAP 2006: 12.091

Session 1: P. ILA

## Classification of elements : Major – Minor -Trace Chemical Variability of the Earth

There are two ideas about igneous rocks that are geologically important.

1. Evolution of igneous rocks - change from one kind of rock into another.
2. Rocks are not randomly distributed across the earth. Specific kinds of rocks are found in specific places due to specific reasons, all connected to plate tectonic processes.

## Chemical Variability of the Earth

**Why is the abundance of different elements varying?**

**What are the effects on human health?**



# Natural & Geological Environment

Natural environment consists of innumerable components, stated simply, consists of

- Atmosphere
- Hydrosphere
- Lithosphere
- Animal life
- Plant life

The components of the natural environment are inter-related closely and interact constantly.

Interferences will cause negative or positive effects on the environment.

Reference: pp 39, Medical geology - Effects of geological environments on human health.

# Basic Components of the Earth

Earth basically consists of Crust, Mantle and Core.

## Crust:

outer most; thinner than the mantle and core; brittle and breakable.

Mantle: below crust;

- ❖ lithosphere - crust and uppermost solid mantle; broken up into the moving plates containing continents and oceans
- ❖ asthenosphere is below the upper mantle  
asthenes – weak  
is hot semi-solid material  
subjected to high temperature and pressure
- ❖ lithosphere is thought to be floating on the asthenosphere

Core :

- ❖ Liquid Outer core
- ❖ Solid Inner core (Fe-Ni alloy)

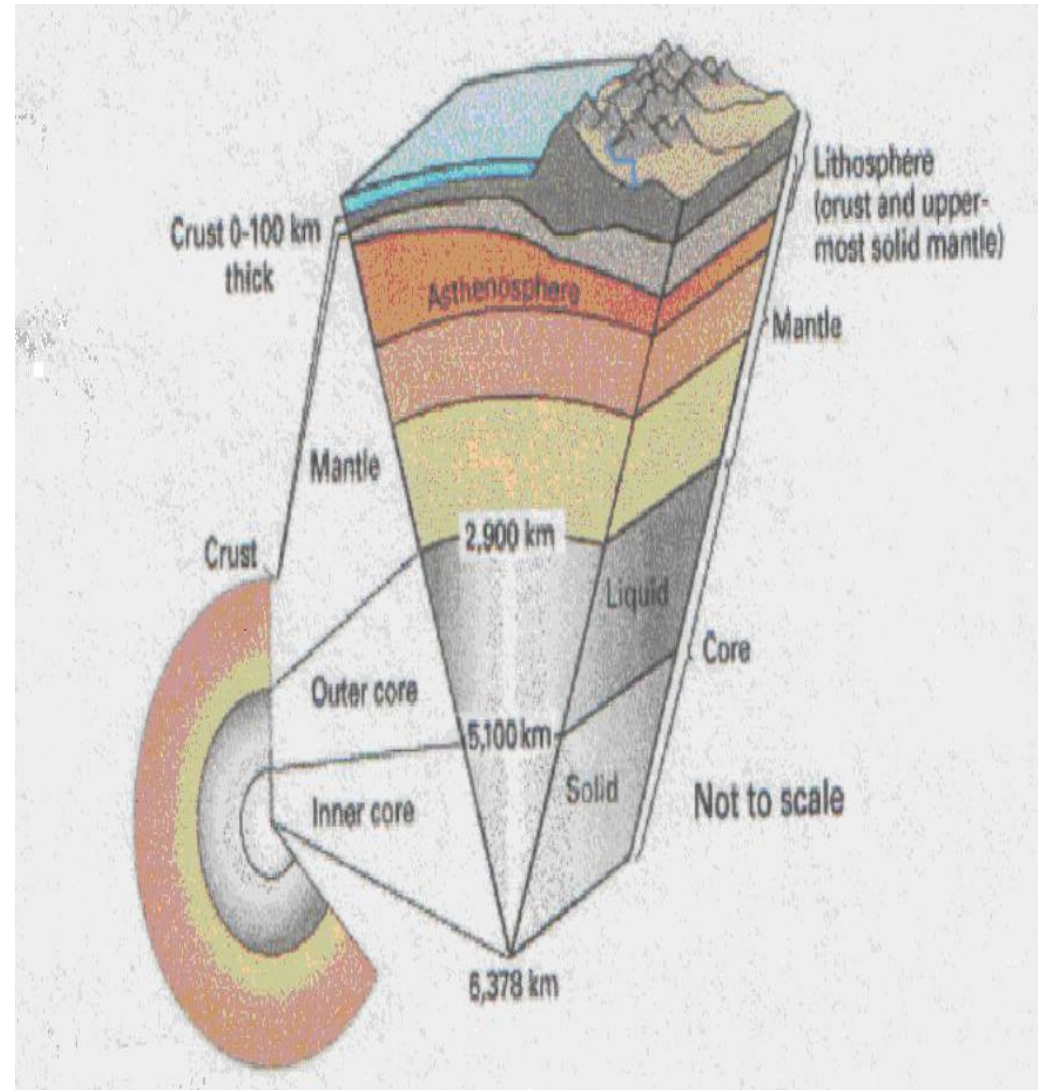


Figure 4. Cross-sectional view of the Earth

Courtesy of USGS.

Reference: <http://pubs.usgs.gov/publications/text/inside.html>.

# Natural & Geological Environment

- For a long time, the understanding was that the relationship between the inner parts of the lithosphere and the biosphere is only mono-directional, i.e., one –way. But the reverse is also true is the current understanding.
- Magmatism and metamorphism:  
Formations of rocks, volcanic exhalations, rock folding, mountain formation and such processes influence the processes like weathering, sediment formation, ground water activity.

# Natural & Geological Environment ...

- A quarter century worth research effort clearly indicates that the reverse influence, namely, the external processes effect the internal processes. For example, processes happening in the biosphere effect the composition of hydrothermal solutions, sedimentary formation of sulphur.
- “Thus, there is unity of the lithosphere as a complex dynamic system with feed back from the biosphere.” – M. Komatina

Reference: pp 45, Medical geology - Effects of geological environments on human health.

# Geological Environment

- **Igneous rocks:**

Rocks formed by the cooling and solidification of hot molten magma. Magma is a material formed by localized melting within the Earth.

- **Intrusive igneous rocks** are formed by If the magma solidification beneath the Earth's surface. Example: Granite.

- **Extrusive igneous or volcanic rocks** are formed by the solidification of magma above Earth's surface in the form of lava. Example: Basalt.

# Geological Environment ...

**Sedimentary rocks** are formed at the Earth's surface:

- **Clastic sedimentary rocks** from the weathered and eroded fragments of pre-existing rocks – example: quartz sandstone;
- **Organic sedimentary rocks** from the hard parts of animals or plants – example: shelly limestone, coal;
- **Chemical sedimentary rocks** from the precipitation out of solution of dissolved minerals – example: rock salt, gypsum.

# Geological Environment ...

The color/composition of the rock is, at its simplest, divided into

- Mafic: dark colored rocks
- Intermediate colored rocks.
- Felsic: light colored rocks.
- Ultramafic: Some characteristics of mafic, but lack some others. Example: Dunite, Peridotite

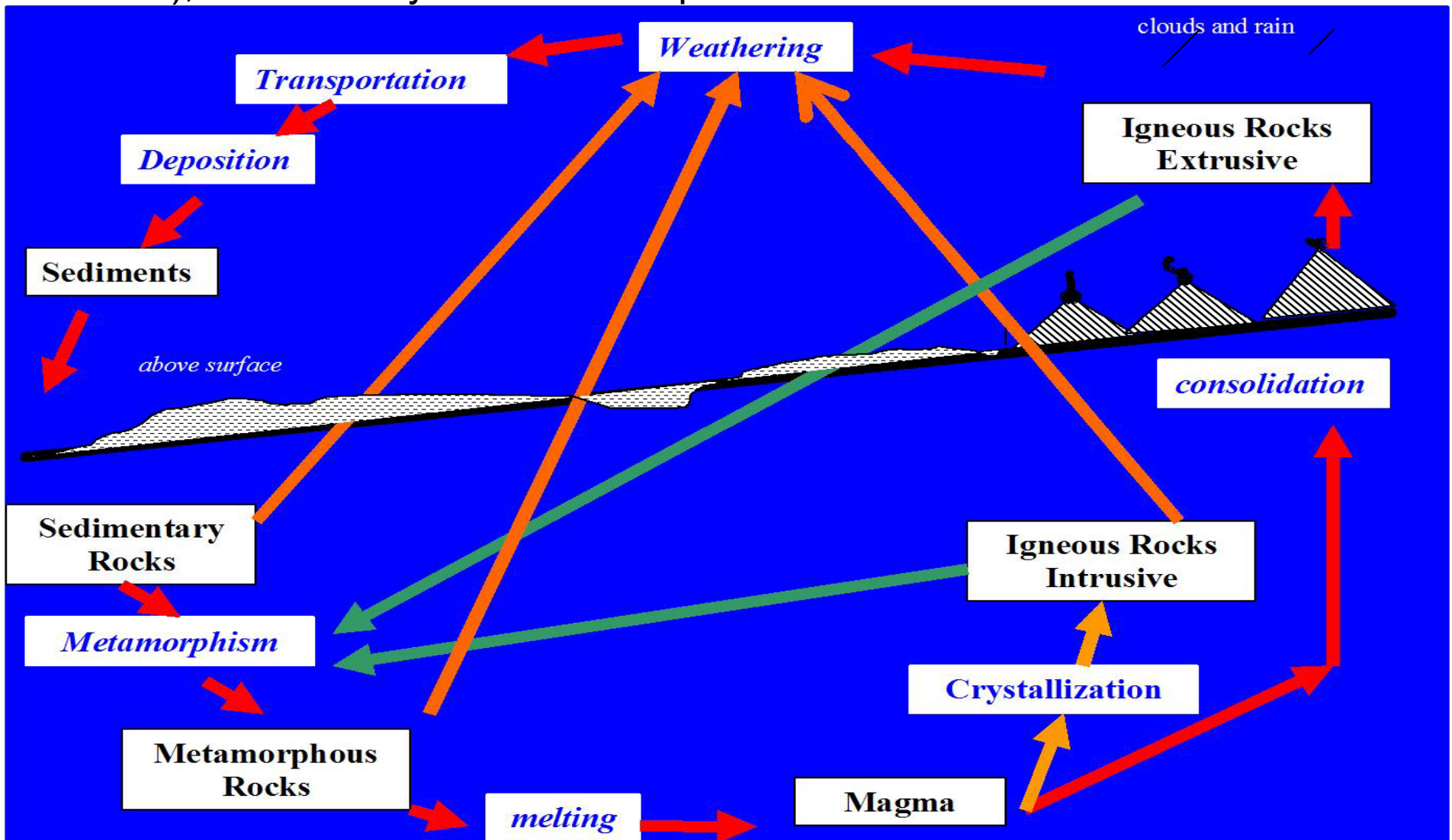
# Geological Environment

- Metamorphic rocks are formed from the mineralogical and/or textural **transformation**, in the solid state, of **pre-existing rocks** due to the **action** of temperature and/or pressure.
- Metamorphic rocks that have been subjected
  - (1) to **deep burial** typically display a foliated texture due to the parallel alignment of some constituent minerals – example: schist;
  - (2) to the **segregation of minerals** into separate bands of different composition, example: gneiss



## Figure 5. Rock Cycle

Earth processes involving the atmosphere, lithosphere, crust and upper mantle and the cycle of three types of rocks, namely, igneous (intrusive and extrusive), sedimentary and metamorphous rocks are all inter-related.



# Natural & Geological Environment ...

- The geochemical associations of different elements are responsible for the formation of different minerals which become Earth's major natural sources.
- Some geochemical associations are listed.
- Geochemists observed consistent patterns in the distribution of many elements.

# Table 4.

## Observation of geochemical associations of elements in some Igneous and Sedimentary rock types

| Description               | Association                   |
|---------------------------|-------------------------------|
| <b>Igneous rocks:</b>     |                               |
| <b>Mafic</b>              | <b>Mg-Fe-Ti-V</b>             |
| <b>Ultramafic</b>         | <b>Mg-Fe-Cr-Ni-Co</b>         |
|                           |                               |
| <b>Sedimentary rocks</b>  |                               |
| <b>Fe-oxide enhanced</b>  | <b>Fe-As-Co-Ni-Se</b>         |
| <b>Mn-oxide enhanced</b>  | <b>Mn-As-Ba-Co-Mo-Ni-V-Zn</b> |
| <b>Phosphate enhanced</b> | <b>P-F-U-Cd-Ag-Pb-Mo</b>      |

Based on Table II, pp 27, Essentials of Medical Geology.

# Natural & Geological Environment ...

## Simple Explanation:

- The composition of the individual minerals that form the rock influences the elemental diversity.
- The properties are carried forward to other materials through processes of
  - erosion**
  - weathering**
  - soil formation**
- And to water that passes through the solid phase materials



# Natural & Geological Environment ...

**What is the link between the elements of the Earth and its environment to the humans and plants?**

# **Elemental link between geosphere and biosphere**

**An attempt to understand**

# Link between the geosphere and the biosphere ...

Earth basically consists of Crust, Mantle and Core.

**Crust:** outer most; thinner than the mantle and core; brittle and breakable

**Mantle:** below crust;

lithosphere - crust and uppermost solid mantle; broken up into the moving plates containing continents and oceans

asthenosphere is below the upper mantle  
asthenes – weak

is hot semi-solid material subjected to high temperature and pressure

lithosphere is thought to be floating on the asthenosphere

**Core :**

Liquid Outer core

Solid Inner core (Fe-Ni alloy)

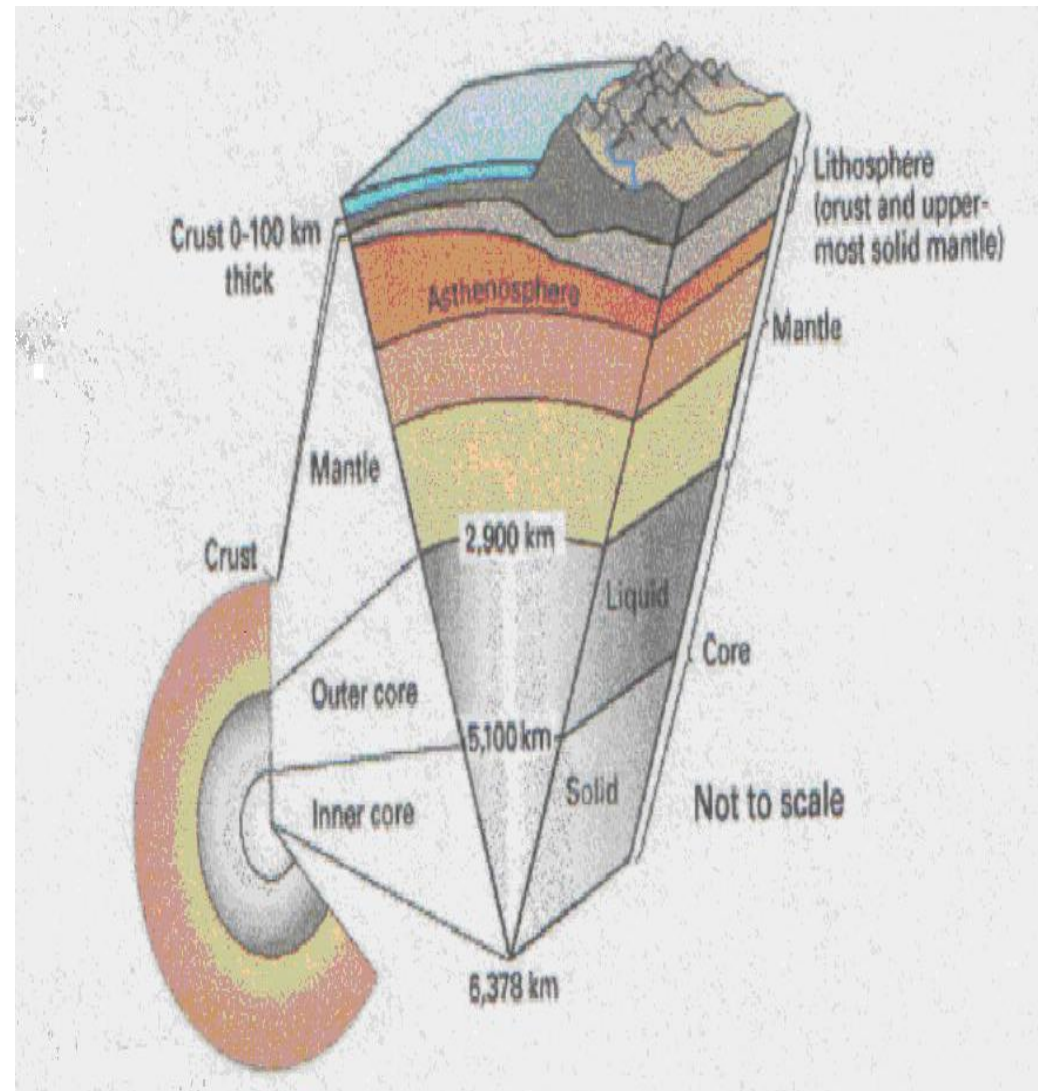


Figure 4. Cross-sectional view of the Earth

Courtesy of USGS.

Reference: <http://pubs.usgs.gov/publications/text/inside.html>.

# Link between the geosphere and the biosphere

In order to visualize the very first link between the biosphere and the geosphere, 4 billion years ago, one has to focus on the essential characteristics of the primitive earth and the primitive cell of the living organism [Reference. R. J. P. Williams].

Based on Chapter 4: Uptake of elements from a chemical point of view, R. J. P. Williams, pp 61- 85, Essentials of Medical Geology.



# Link between the geosphere and the biosphere ...

## Change of oxidation states of elements with increased oxygen with time

Prof. R. J. P. Williams explained excellently, in his article “Uptake of elements from a chemical point of view”:

- the influence of primitive sea on the chemical elements of chemical system, which in turn effected the primitive life,
- the free elements of the primitive sea self generated a chemical system of interactions that could be seen in cytoplasm of all cells even in today,
- the evolution of the free and bound the metallomes and metabollomes was largely due to the formation of large compartments, and in oxidative possibilities

Note: Metallome, coined by R.J.P. Williams, means distribution of free metal ions in every one of cellular compartments.

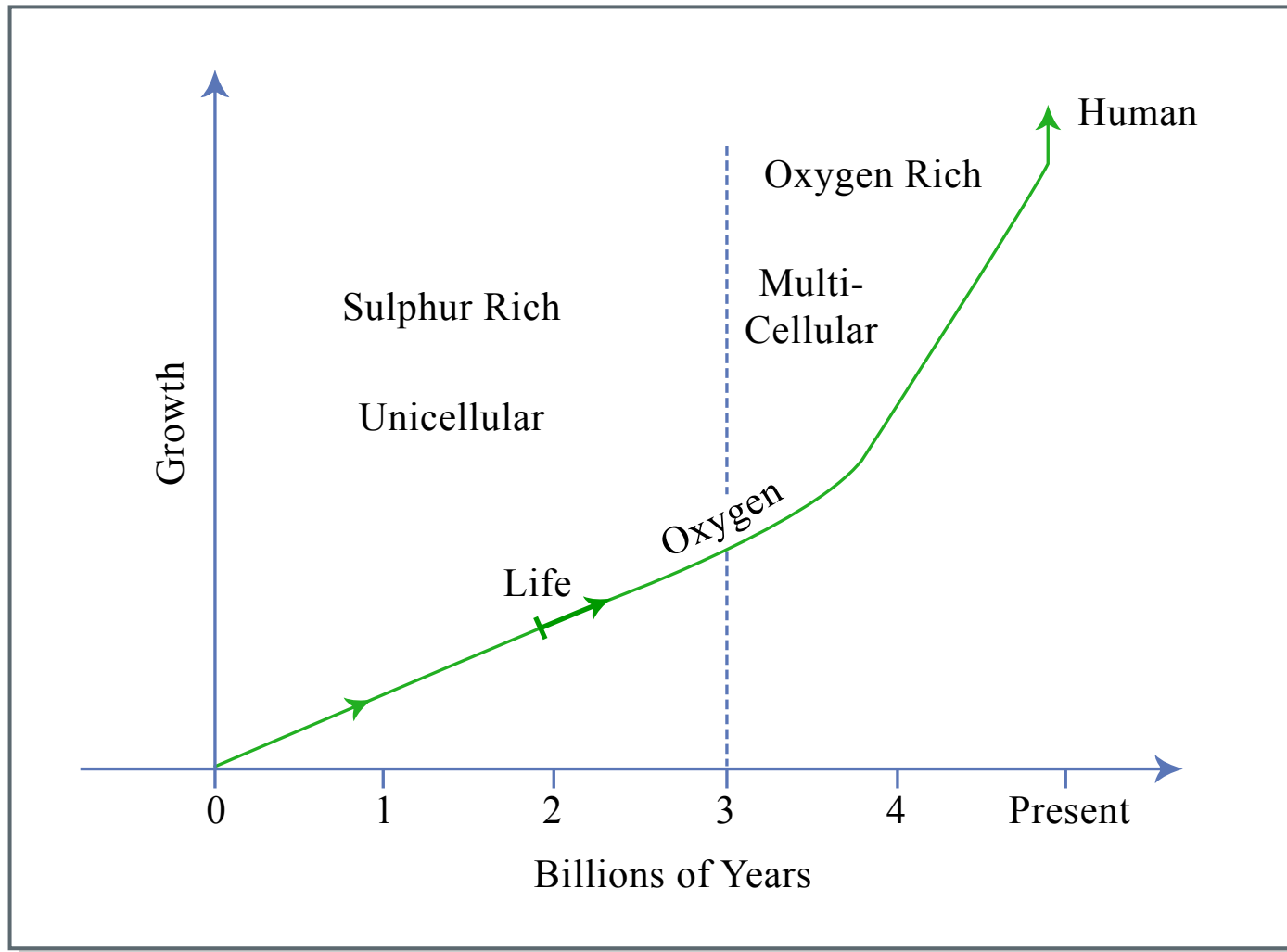
Metabollome (or metabolome) means the entire set of metabolic entities and small network pathways in a cell, tissue, organ, organisms, and species.

Ref. Chemical selection of elements by cells, R.J.P. Williams, Coordination Chemistry Reviews 216–217: 583–595;2001

Chapter 4: Uptake of elements from a chemical point of view, R. J. P. Williams, pp 61- 85, Essentials of Medical Geology.

# Link between the geosphere and the biosphere ...

Figure 6. Oxygen and Cellular changes with time



Based on Figure 11, pp 77, Essentials of Medical Geology.

Figure by MIT OCW.

# Table 5. Time sequence of geological and biological changes

| Time              | Biological  | Interpretation  | Oxygen | Elements | Elements | Elements      |
|-------------------|---|---|--------|----------|----------|---------------|
| Million years ago | Evidence  |   | %      | Loss     | Gain     | Little change |
| 400               | Large fishes, first land plants                         |   | 100    | (1)      | (2)      |               |
| 550               | Cambrian fauna  | Shelly metazoans<br>Absorption through external shell | 10     | ↑        | ↑        | ↑             |
| 670               | Ediacarian fauna  | Metazoans, collagen                                   | 1      |          |          |               |
| 1400              | Cells larger  | Eukaryotic cells                                      | >1     | ↑        | ↑        | (3)           |
| 2000              | Enlarged thick walled cells                             | Oxygen tolerating blue green algae                    | 1      |          |          |               |
| 2800              | Filamentous Chains, Stromatolites                       | Resemble blue green algae                             | 0.1    |          |          | ↓             |
| > 3500            | Stromatolites, depletion of <sup>13</sup> C             | Precursors of blue green algae                        | <0.01  |          |          |               |
| 3800              | Rhythmically banded rocks, depletion of <sup>13</sup> C | Microbial organisms (?)                               | <0.01  |          |          |               |

- (1) Fe<sup>2+</sup>, S<sup>2-</sup>, Se<sup>2-</sup>, H<sub>2</sub>, MoS<sub>4</sub><sup>2-</sup>, NH<sub>3</sub>, CO<sub>2</sub>  
 (2) Cu<sup>2+</sup>, Zn<sup>2+</sup>, Cd<sup>2+</sup>, Fe<sup>3+</sup>, MoO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, SeO<sub>4</sub><sup>2-</sup>, I<sub>2</sub>  
 (3) Mn<sup>2+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Si(OH)<sub>4</sub>, HPO<sub>4</sub><sup>2-</sup>, Cl<sup>-</sup>, Na<sup>+</sup>, K<sup>+</sup>

Based on Figure 11, pp 77, Essentials of Medical Geology.

## Chemical Variability of the Earth

**Why is the abundance of different elements varying?**

# Essential and Non-essential Elements

## With Reference to Human Health

Introduction

Essential vs Non-essential

Sources of Exposure

Health Effects

Metal Induced Changes

Toxicity

Carcinogenesis

Summary

# Essential and Non-Essential Elements with reference to human health

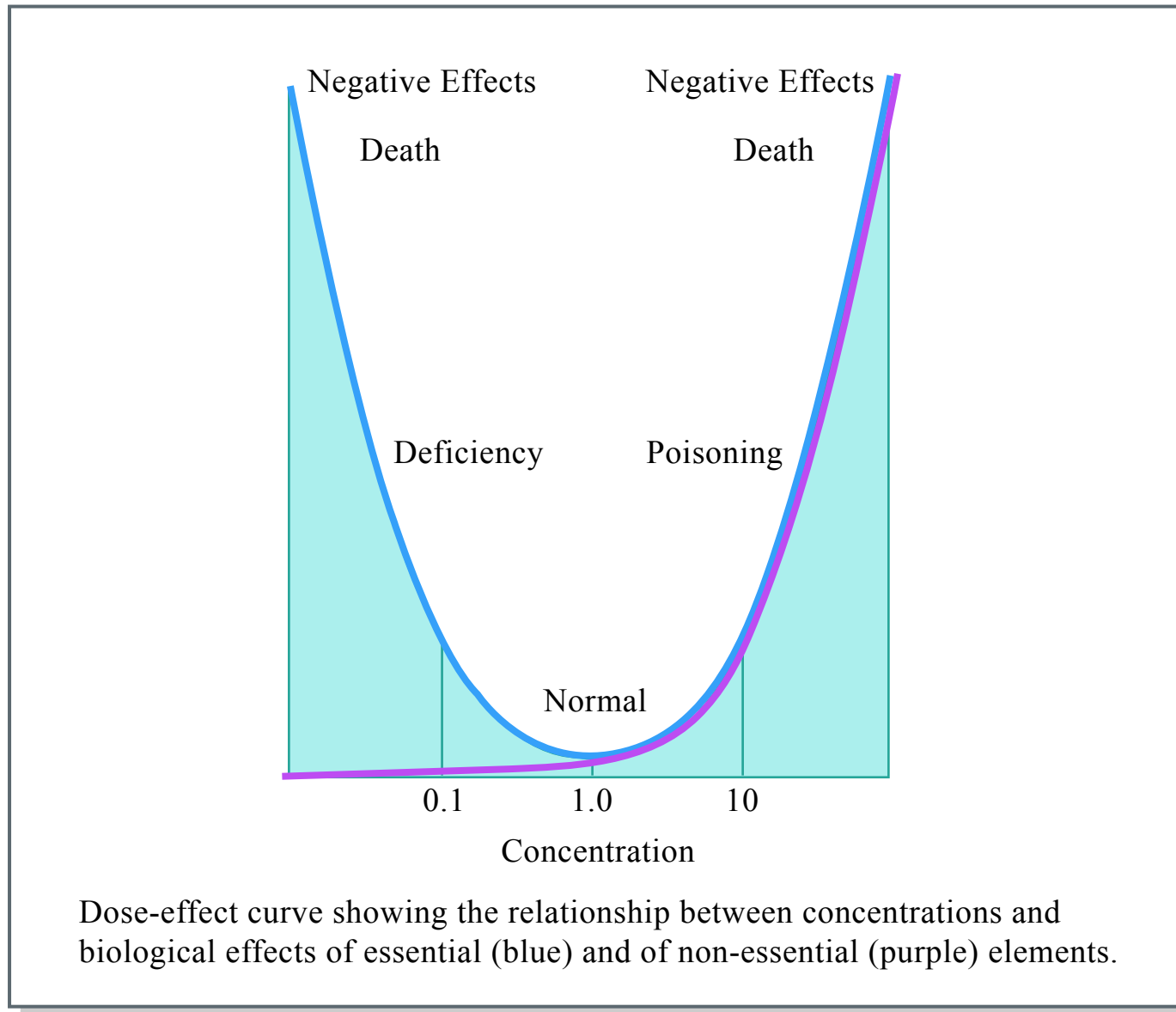
## Introduction:

### Essential element considerations in human health:

- Presence in healthy tissues
- Constant concentrations
- Withdrawal from the system causes reproducible physiological symptoms (abnormalities)
- Addition should reverse the conditions
- Physiological changes, due to excess or deficient concentrations of element (or elements), should be preventable or even be cured when the excess or deficient condition of the element(s) is prevented or cured.

Based on pp 88, Essentials of Medical Geology.

# Figure 7. Dose Response – Health Effects



## Table 6.

### Major - Minor – Trace elements of human body

| Concentration | Element                                | Classification |
|---------------|--|----------------|
| 3 - 65 %      | C, H, N, O                             | Major elements |
| 0.1 - 1%      | Ca, Cl, K, Mg, Na, P, S (Electrolytes) | Minor elements |
| <100mg/kg     | As, Br, Co, Cr, Cu, F, Fe, I, Li       | Trace elements |
|               | (All are not listed)                   |                |



# Table 7.

## Major-Minor Elements of Human Body

| Concentration | Element | Classification |
|---------------|---------|----------------|
| 65%           | O       | Major          |
| 18%           | C       | Major          |
| 10%           | H       | Major          |
| 3.00%         | N       | Major          |
| 1.40%         | Ca      | Minor          |
| 0.50%         | Mg      | Minor          |
| 0.34%         | K       | Minor          |
| 0.26%         | S       | Minor          |
| 0.14%         | Na      | Minor          |
| 0.14%         | Cl      | Minor          |

Based on Table 1, pp 116, Essentials of Medical Geology.

## Table 8. Trace Elements of Human Body

| Concentration | Element | Classification |
|---------------|---------|----------------|
| 0.26 mg/kg    | As      | Trace          |
| 2.9 mg/kg     | Br      | Trace          |
| 0.021 mg/kg   | Co      | Trace          |
| 0.094mg/kg    | Cr      | Trace          |
| 1 mg/kg       | Cu      | Trace          |
| 37 mg/kg      | F       | Trace          |
| 60 mg/kg      | Fe      | Trace          |
| 0.019 mg/kg   | I       | Trace          |
| 0.009 mg/kg   | Li      | Trace          |
| 0.17 mg/kg    | Mn      | Trace          |
| 0.08 mg/kg    | Mo      | Trace          |
| 0.14 mg/kg    | Ni      | Trace          |
| 0.11 mg/kg    | Se      | Trace          |
| 260 mg/kg     | Si      | Trace          |
| 0.24 mg/kg    | Sn      | Trace          |
| 0.11 mg/kg    | V       | Trace          |
| 0.008 mg/kg   | W       | Trace          |
| 33 mg/kg      | Zn      | Trace          |

Based on Table II, pp116, Essentials of Medical Geology.

# Pathways of Metals in Environment

## Effects on Health

- Pathways:

  - Air (inhalation)

  - Absorption

  - Drinking water

  - Food cycles

- Metal induced effects

  - Carcinogenic, Teratogenic, Mutagenic

Teratogenic: mis-shapen animal or plant.

# Trace Element Deficiency & Toxicity Health Effects

- Established Toxic Elements are
  - Arsenic
  - Cadmium
  - Lead
  - Mercury
  - Radon

**Table 9.**  
**Toxicity: Cancer Effects**

| <b>Elements</b> | <b>Target organ/site</b>  |
|-----------------|---|
| As              | Bladder, liver, lung, skin,<br>Vascular and neurological changes          |
| Cd              | Bladder, kidney, lung,<br>Hypertension, vascular and neurological changes |
| Cr              | Kidney, liver, lung   |
| Hg              | Brain, kidney, lung, neurological changes                                 |
| Pb              | Neurological, IQ (children), Anemia                                       |
| Radon           | Lung  |
| Others          | Be (skin, lung), Co (liver, lung), U (skin)                               |

# Table 10.

## Toxicity: Pigmentation disorders

| <b>Elements</b> | <b>Effect</b>               |
|-----------------|-----------------------------|
| Arsenic         | Hyper and hypo pigmentation |
| Gold            | Chrysiasis                  |
| Mercury         | Hyper pigmentation          |
| Silver          | Argyria                     |

Based on: Trace elements in environmental health and human diseases, Josè- Diversity of Trace Elements.pdf

# Table 11.

## Toxicity: Arsenic induced effects

**Pigmentation (hyper and hypo)**

**Keratosiis**

**Bowen's disease**

**Squamous cell carcinoma**

**Basal cell carcinoma**

Based on: Trace elements in environmental health and human diseases, Josè- Diversity of Trace Elements.pdf

# Table 12.

## Toxicity: Cardiovascular diseases

| Elements                       | Effect          |
|--------------------------------|-----------------|
| ❖ From epidemiological studies |                 |
| ❖ Co, Hg, Se, Al, As, Au, Cr   | Cardiomyopathy  |
| ❖ Fe, Se, Ca, Cu, Mg           | Atherosclerosis |
| Al, As, Hg, ❖ Pb               | Hypertension    |

Based on: Trace elements in environmental health and human diseases, Josè- Diversity of Trace Elements.pdf

Atherosclerosis: Hardening and thickening of walls of arteries with fatty degeneration

Armed forces

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# Table 13.

## Toxicity: Lung diseases

| <b>Elements</b>                      | <b>Effect</b>   |
|--------------------------------------|---|
| Asbestos fibers                      | Amphipbole types are more pathogenic than serpentine type |
| Beryllium                            | Presence of granulomas                                    |
| Iron;<br>Iron mixed with free silica | Siderosis;<br>silicosiderosis                             |

Based on: Trace elements in environmental health and human diseases, Josè- Diversity of Trace Elements.pdf

# Table 14.

## Toxicity: Liver diseases *(Hepatotoxicity)*

| Element   |
|-----------|
| Aluminium |
| ❖ Arsenic |
| Barium    |
| Beryllium |
| Cadmium   |
| Chromium  |
| ❖ Copper  |
| Gold      |
| ❖ Iron    |
| Lead      |

❖ From epidemiological studies

Based on: Trace elements in environmental health and human diseases, Josè- Diversity of Trace Elements.pdf

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# Table 15.

## Toxicity: Kidney diseases *(Nephrototoxicity)*

### Element

❖ Arsenic

Bismuth

❖ Cadmium

❖ Chromium

Gold

❖ Lead

Lithium

Platinum

❖ From epidemiological studies

## Table 16: Significant Endemic Diseases

| Element  | Effect  |
|----------|---|
| Arsenic  | Skin lesions, Cancer  |
| Fluoride | Dental and Skeletal   |
| Iodine   | Goiter and Cretinism  |
| Selenium | Kaschin-Beck disease:<br><i>Degenarative osteoarthropathic disease</i><br>Keshan disease: <i>Chronic heart disease (cardiomyopathy)</i> |

Based on: Trace elements in environmental health and human diseases, Josè- Diversity of Trace Elements.pdf

**Table 17.****Diseases due to deficiency and toxicity of some elements**

| <b>Element</b>   | <b>Deficiency</b>                                   | <b>Toxicity</b>                  |
|------------------|---|----------------------------------|
| <b>Chromium</b>  | <b>Disturbances in the glucose metabolism</b>       | <b>Kidney damage (Nephritis)</b> |
| <b>Cobalt</b>    | <b>Anemia , “White Liver disease”</b>               | <b>Heart failure</b>             |
| <b>Copper</b>    | <b>Anemia, poor growth, bone decreased in WBC</b>   | <b>Idiopathic Cu toxicosis</b>   |
| <b>Iron</b>      | <b>Anemia</b>                                       | <b>Hemochromatosis</b>           |
| <b>Magnesium</b> | <b>Convulsions, malfunctions of the skeleton</b>    |                                  |
| <b>Selenium</b>  | <b>Liver necrosis</b>                               | <b>Muscular dystrophy</b>        |
| <b>Zinc</b>      | <b>Dwarf growth, retarded development of gonads</b> | <b>“Metallic” fever</b>          |

Based on: Trace elements in environmental health and human diseases, Josè- Diversity of Trace Elements.pdf

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# Summary

- Trace element deficiency or excess (toxicity) cause a wide range of environmentally related health problems.
- Toxicity may effect more than one organ in the system.
- Toxicity may vary with chemical or physical form of the metals.
- Deficiency related effects may be treatable (by providing dietary supplements etc.)

# Selection of Elements for Report and Presentation

Format:

Introduction

Experimental Method

Facts

Results and Conclusion

References



# Summary

I talked about

- Different Terminologies of Medical Geography, Geomedicine, Medical Geology/Geochemistry
- Classification of Elements
  - Periodic Table of Elements
  - Major-Minor-Trace Elements
  - Geochemical Empirical Classifications
  - Essential and Non-essential Elements
- Link between the geosphere and biosphere





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# Summary ...

I talked about

- Essentiality and non-essentiality of elements with reference to human health.
- Sources of exposure of toxic elements.
- Health effects induced by deficiency and excess of metals.



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# Internet Keywords

- medical geology, medical geochemistry
- major, minor, trace elements
- essential and nonessential elements
- periodic table of elements
- toxic elements
- Primitive earth, primitive sea



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