

Zeeman Effect

Read CTDL, pp. 1156-1178, preparation for 1-D solids

LAST TIME:

- * computational tricks and inter-relationships among atomic and molecular constants
- * $\mathbf{H}^{\text{SO}} = \zeta(\text{N,L,S})\mathbf{L}\cdot\mathbf{S} \rightarrow$ a different $\zeta(\text{N,L,S})$ constant for each L-S state in configuration N

or, from $\mathbf{H}^{\text{SO}} = \sum_i a(r_i)\ell_i s_i \rightarrow$ one $\zeta_{n\ell}$ parameter valid for all diagonal and off-diagonal intra-configurational matrix elements

$\zeta(\text{N,L,S})$ a fit parameter

$\zeta_{n\ell}$ an orbital integral

micro \rightarrow macro forms of operators

TODAY:

1. electrons \leftrightarrow holes
shortcuts for e^2/r_{ij} and \mathbf{H}^{SO}
2. Hund's 3rd Rule: the lowest L-S-J level of an electronic configuration
3. Zeeman effect. Landé g-factor via Wigner-Eckart Theorem.
Another way to evaluate reduced matrix elements.
4. $\mathbf{H}^{\text{Zeeman}}$ in Slater determinantal basis set.

5.73 Lecture #36

36 - 2

1. electrons (e^-) ↔ holes (h^+) shortcuts

subshell	$n\ell^N$			
½ full	s	p	d	f
# e^-	1	3	5	7

for p^5 do we need to consider $5e^-$ or one h^+ ?

$$p^5 \quad {}^2P \quad M_L = 1, M_S = 1/2 \quad ||1\alpha 1\beta 0\alpha 0\beta - 1\alpha||$$

$$(ph^+)^1 {}^2P \quad M_L = 1, M_S = 1/2 \quad ||1\alpha||$$

Get same energy order of L-S states via $5e^-$ or $1h^+$ Slater determinant

Why? e^2/r_{ij} expresses $e^- - e^-$ repulsion
 $h^+ - h^+$ is also repulsion

so we get the same result for e^2/r_{ij} by e^- or h^+ algebra. All F_k, G_k Slater-Condon parameters are > 0 .

What about p^5d^1 vs. p^1d^1 ? Is the energy level order inverted? Hund's 1st and 2nd rules predict the same lowest state in p^5d as in $(ph^+)^1 d^1$.

What about e^- vs. h^+ for H^{SO} ?

$$H^{SO} = \zeta(N, L, S) \mathbf{L} \cdot \mathbf{S} = \sum_i a(r_i) \ell_i \cdot \mathbf{s}_i$$

Take $\Delta M_J = 0$, $M_J = J = L + S$ matrix element of both sides of the H^{SO} equation.

This is an extreme state from the M_L, M_S box diagram. There is only one Slater determinant in this box.

5.73 Lecture #36

36 - 3

$p^5 \ ^2P$ vs. $p^1 \ ^2P$

$$\begin{aligned} \hbar^2 \zeta(p^5 \ ^2P, M_L = 1, M_s = 1/2) \left(1 \cdot \frac{1}{2}\right) &= \hbar^2 \zeta_{np} \left[1 \cdot \frac{1}{2} + 1 \cdot \left(-\frac{1}{2}\right) + 0 + 0 + -1 \cdot \frac{1}{2}\right] \\ &= \hbar^2 \zeta_{np} (-1) \left(\frac{1}{2}\right) \end{aligned}$$

$$\zeta(p^5 \ ^2P) = -\zeta_{np}$$

Now $p^1 \ ^2P$

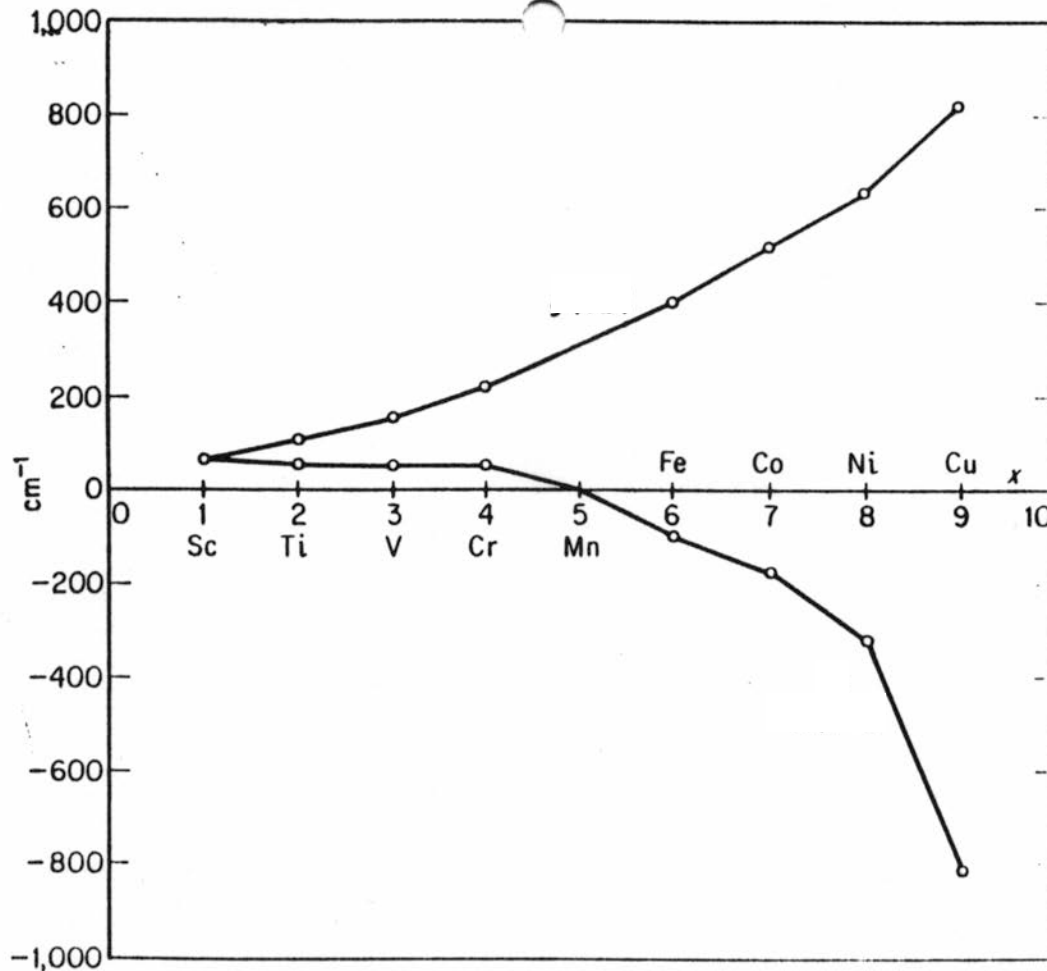
$$\hbar^2 \zeta(p^1 \ ^2P, M_L = 1, M_s = 1/2) \underbrace{L \cdot S}_{1 \cdot \frac{1}{2}} = \hbar^2 \zeta_{np} \left[1 \cdot \frac{1}{2}\right]$$

$$\zeta(p^1 \ ^2P) = \zeta_{np}$$

So we get a sign reversal in $\zeta(N \ ^2P)$ for p^5 vs. p^1 .

So $e^- \leftrightarrow e^+$ gives no change in e^2/r_{ij} expressed by an always positive F_k and G_k . However, get sign reversed in \mathbf{H}^{SO} for $\zeta(N, L, S)$ but not for $\zeta_{n\ell}$.

Figure 6-2 from Tinkham (page 187) for L_{MAX}, S_{MAX} state of $3d^N 4s^2$ configuration



This figure is in the public domain.

Fig. 6-2. Spin-orbit parameters in $3d^x$ transition elements. The splitting parameters $\zeta(LS)$ are computed as indicated in Table 6-2 and averaged over the various splittings. The data used are for the $3d^x 4s^2$ configurations of the neutral atoms. (From Charlotte E. Moore, "Atomic Energy Levels," Natl. Bur. Standards, Circ. 467, vols. I and II, 1949 and 1952. A very similar figure appears in Condon and Shortley.)

Note that ζ_{nl} is a one- e^- orbital integral and is expected to exhibit "periodic" variation. It increases with atomic number and this is expected for increasing Z^{eff} :

$$Z \rightarrow Z + 1, Z^{\text{eff}} \rightarrow Z^{\text{eff}} + 1 - 1/2$$

See Burns' rule paper on shielding included here: *J. Chem. Phys.* **41**, 1521 (1964).

General result for $\zeta(N, L, S)$:

$$\zeta(N, L, S) = \pm \frac{1}{2S_{\text{MAX}}} \zeta_{nl}$$

+ for $n < 2l + 1$

- for $n > 2l + 1$

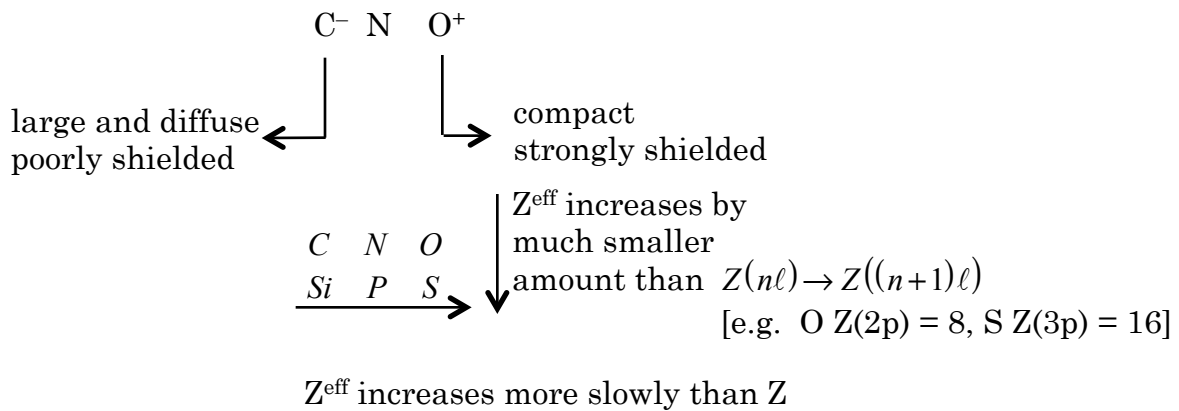
$$\zeta(N, L, S) = 0 \text{ for } n = 2l + 1$$

Burn's paper on Pages 5 and 6 has been removed due to copyright restrictions. See Burn, Gerald. "Atomic Shielding Parameters." *J. Chem. Phys.* **41** (1964): 1521–1522.

5.73 Lecture #36

36 - 7

Isoelectronic species (5 valence electrons):



Hund's 3rd Rule

1st and 2nd Hund's rules (for most stable L - S term) specify S_{MAX} and then L_{MAX} among all of the L - S states of a configuration.

$L_{\text{MAX}} M_{L_{\text{MAX}}} S_{\text{MAX}} M_{S_{\text{MAX}}}$ is a single Slater determinant. It is also equal to

$$J = L_{\text{MAX}} + S_{\text{MAX}} M_{J_{\text{MAX}}} = L_{\text{MAX}} + S_{\text{MAX}}$$

$$\zeta(n\ell^N, L_{\text{MAX}}, S_{\text{MAX}})_{L_{\text{MAX}}, S_{\text{MAX}}} = \zeta_{n\ell} \sum_i m_{\ell_i} m_{s_i}$$

\uparrow
 M_L

\uparrow
 M_S

$$\zeta(n\ell^N, L_{\text{MAX}}, S_{\text{MAX}}) = \zeta_{n\ell} \frac{\sum_i m_{\ell_i} m_{s_i}}{L_{\text{MAX}} S_{\text{MAX}}}$$

Shell $< \frac{1}{2}$ full $N < 2\ell + 1$

All spins are α

$$S_{\text{MAX}} = N / 2$$

$$M_{L_{\text{MAX}}} = \ell + (\ell - 1) + \dots + (\ell - N + 1)$$

Sum includes sum of one ℓ for each of N terms, each additive term decreased by 0, 1, ... $N - 1$.

$$M_{L_{\text{MAX}}} = N(\ell - (N - 1)/2)$$

$$\begin{aligned}
 \zeta(n\ell^N, L_{\text{MAX}}, S_{\text{MAX}}) &= \zeta_{nl} \frac{\frac{1}{2} \sum_i m_{\ell_i}}{L_{\text{MAX}} S_{\text{MAX}}} \quad \begin{array}{l} m_s = \alpha \text{ for all} \\ \text{occupied spin-orbitals} \end{array} \\
 &= \zeta_{nl} \frac{\frac{1}{2} L_{\text{MAX}}}{L_{\text{MAX}} (N/2)} = \zeta_{nl} / N = \zeta_{nl} / 2S_{\text{MAX}} \quad \begin{array}{l} M_L = L_{\text{MAX}} \end{array}
 \end{aligned}$$

Similarly for shell that is more than half-full.

$$\begin{aligned}
 S_{\text{MAX}} &= \frac{1}{2} \{ (2\ell+1)\alpha \text{ spins} - [N - (2\ell+1)]\beta \text{ spins} \} \\
 &= (2\ell+1) - N/2 \\
 M_S = S_{\text{MAX}} &= (2\ell+1) - N/2
 \end{aligned}$$

$$\begin{aligned}
 \zeta(n\ell^N, L_{\text{MAX}}, S_{\text{MAX}}) &= \zeta_{nl} \frac{\left[\frac{1}{2} \sum_{(\alpha)} m_{\ell_i} - \frac{1}{2} \sum_{(\beta)} m_{\ell_i} \right]}{L_{\text{MAX}} S_{\text{MAX}}} \quad \begin{array}{l} = 0 \\ L_{\text{MAX}} \end{array} \\
 &= -\frac{\zeta_{nl}}{2S_{\text{MAX}}} = -\frac{\zeta_{nl}}{(2\ell+1) - N/2}
 \end{aligned}$$

5.73 Lecture #36

36 - 9

So we have

$$\zeta(n\ell^N, L_{\text{MAX}}, S_{\text{MAX}}) = \frac{\zeta_{nl}}{2S_{\text{MAX}}} \quad N < 2\ell + 1$$

$$S_{\text{MAX}} = N/2 = \frac{1}{2} \# \text{ of } e^- \quad N = 2\ell + 1$$

$$= -\frac{\zeta_{nl}}{2S_{\text{MAX}}} \quad N > 2\ell + 1$$

$$S_{\text{MAX}} = [(2\ell + 1) - N/2] = \frac{1}{2} (\# \text{ of } h^+)$$

Shell half full

$$\left. \begin{array}{l} L_{\text{MAX}} = 0 \\ S_{\text{MAX}} = N/2 \end{array} \right]^{N+1} S_{N/2}$$

No fine structure. Only a single J-Level, $J = N/2$.

3. Zeeman Effect.

Provides additional information for assignment of an L-S-J state.

$$\mathbf{H}^{\text{Zeeman}} = - \left(\underbrace{\mu_0 / \hbar}_{\substack{\text{Bohr Magneton} \\ 1.399613 \text{ MHz/Gauss} \\ 1 \text{ cm}^{-1} = 29979 \text{ MHz}}} \right) \left(L_Z + 2S_Z \right) \underbrace{B_Z}_{\substack{\text{magnetic field} \\ \text{(entirely along Z} \\ \text{laboratory axis)}}}$$

In the coupled representation, $|JM_JL_S\rangle$ matrix elements of $\mathbf{H}^{\text{Zeeman}}$ are awkward because it is necessary to transform from coupled to uncoupled $|LM_LSM_S\rangle$ basis set in order to evaluate matrix elements of $\mathbf{H}^{\text{Zeeman}}$.

This is a fast way to get a closed-form expression for matrix elements of $\mathbf{H}^{\text{Zeeman}}$ in the coupled basis set. It involves a clever way to evaluate reduced matrix elements using the Wigner-Eckart Theorem.

\mathbf{H}^{SO} and e^2/r_{ij} are rigorously diagonal in J (WHY?) but $\mathbf{H}^{\text{Zeeman}}$ is not (WHY?). [Answers depend on how you classify an operator relative to \vec{J} .]

We have a battle between [\mathbf{H}^{SO} and e^2/r_{ij}], which is diagonal in J, thus it tries to *defend* the coupled representation against $\mathbf{H}^{\text{Zeeman}}$, which is non-diagonal in J and tries to *destroy* the coupled representation.

We are interested in the weak field limit where

$$\langle J' | \mathbf{H}^{\text{Zeeman}} | J \rangle \ll |E_{J'}^{(0)} - E_J^{(0)}| \text{ due to the operators } \mathbf{H}^{\text{SO}} \text{ and } e^2 / r_{ij}$$

Our special case, limited to $\Delta J = 0$ matrix elements, fails when ζ (NLS) is small and B_Z is large.

Wigner Eckart Theorem tells us

$$\langle JM'LS | \mathbf{L} | JMLS \rangle = \langle JLS | \|\mathbf{L}\| | JLS \rangle \langle JM'LS | \mathbf{J} | JM_jLS \rangle$$

Similarly for \mathbf{S}

$$\langle JM'LS | \mathbf{S} | JMLS \rangle = \langle JLS | \|\mathbf{S}\| | JLS \rangle \langle JM'LS | \mathbf{J} | JMLS \rangle$$

but $J = L + S$

$$\langle \mathbf{J} \rangle = \left(\underbrace{\langle \|\mathbf{L}\| \rangle}_{(1-\alpha)} + \langle \|\mathbf{S}\| \rangle_{\alpha} \right) \langle JM'LS | \mathbf{J} | JMLS \rangle$$

=1

Thus $\langle \mathbf{S} \rangle = \alpha \langle \mathbf{J} \rangle$ or $\boxed{\mathbf{S} = \alpha \mathbf{J}}$! An operator replacement!

Now for the cleverness:

$$\mathbf{J} = \mathbf{L} + \mathbf{S}$$

$$\mathbf{J} - \mathbf{S} = \mathbf{L}$$

$$(\mathbf{J} - \mathbf{S})^2 = \mathbf{L}^2$$

$$\mathbf{J}^2 + \mathbf{S}^2 - 2\mathbf{J} \cdot \mathbf{S} = \mathbf{L}^2$$

$$\mathbf{J} \cdot \mathbf{S} = \frac{\mathbf{J}^2 + \mathbf{S}^2 - \mathbf{L}^2}{2}$$

Matrix elements of \mathbf{J}^2 , \mathbf{S}^2 , and \mathbf{L}^2 are explicitly evaluable in the coupled representation, $|JLSM_j\rangle$.

5.73 Lecture #36

36 - 12

Thus:

$$\langle JM'LS | \mathbf{J} \cdot \mathbf{S} | JMLS \rangle = \delta_{M'M} \hbar^2 \left[\frac{J(J+1) + S(S+1) - L(L+1)}{2} \right]$$

but $\mathbf{J} \cdot \mathbf{S} = \alpha \mathbf{J}^2$

$$\langle JM'LS | \alpha \mathbf{J}^2 | JMLS \rangle = \delta_{m'm} [\hbar^2 \alpha J(J+1)]$$

we can solve for α

$$\alpha J(J+1) = \frac{J(J+1) + S(S+1) - L(L+1)}{2}$$

$$\boxed{\alpha = \frac{J(J+1) + S(S+1) - L(L+1)}{2J(J+1)}}$$

$$\mathbf{H}^{\text{Zeeman}} - \mu_0 B_z (\mathbf{L}_z + 2\mathbf{S}_z) = -\mu_0 B_z (1 + \alpha) \mathbf{J}_z$$

$\downarrow \qquad \qquad \downarrow$
 $(1-\alpha)\mathbf{J}_z \qquad \alpha\mathbf{J}_z$

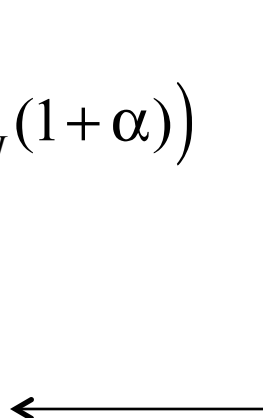
$$\langle JMLS | \mathbf{H}^{\text{Zeeman}} | JMLS \rangle = -\mu_0 B_z \hbar M_J \underbrace{(1 + \alpha)}_{g_J}$$

Zeeman energy level tuning rate:

$$\frac{dE}{dB_z} = \frac{d}{dB_z} \left(-\mu_0 B_z M_J (1 + \alpha) \right)$$

$$= -\mu_0 M_J g_J$$

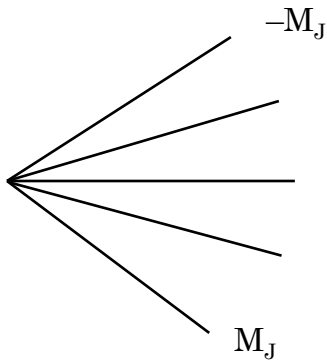
$$g_J = -\frac{1}{\mu_0} \frac{1}{M_J} \frac{dE}{dB_z}$$



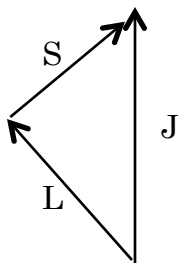
5.73 Lecture #36

Apply a magnetic field, each J level splits into $2J + 1$ M_J components.

of components tells us J, g_J tells us something about L and S



tuning is linear in B_z



$$J_{MAX} = L + S$$

L and S parallel (magnetic moments add)

$$J_{MIN} = |L - S|$$

L and S anti-parallel (magnetic moments partially cancel)

e.g.

for J = 3	L = 0, S = 3	L = 1, S = 2	L = 2, S = 1	L = 3, S = 0
g_J	2.000	1.667	1.333	1.000
or for L = 3, S = 1		J = 4	J = 3	J = 2
g_J		1.250 parallel	1.0833	0.667 anti-parallel

5.73 Lecture #36

36 - 14

To determine J: apply B_Z field and count M_J components

light Z polarized, $\Delta M_J = 0$

J' set of equal $M'_J, M'_J - 1$ splittings
 J'' another set of equal $M''_J, M''_J - 1$ splittings
 $\left. \vphantom{\begin{matrix} J' \\ J'' \end{matrix}} \right\} g_{J'} \neq g_{J''}$

(' for upper state, '' for lower state)

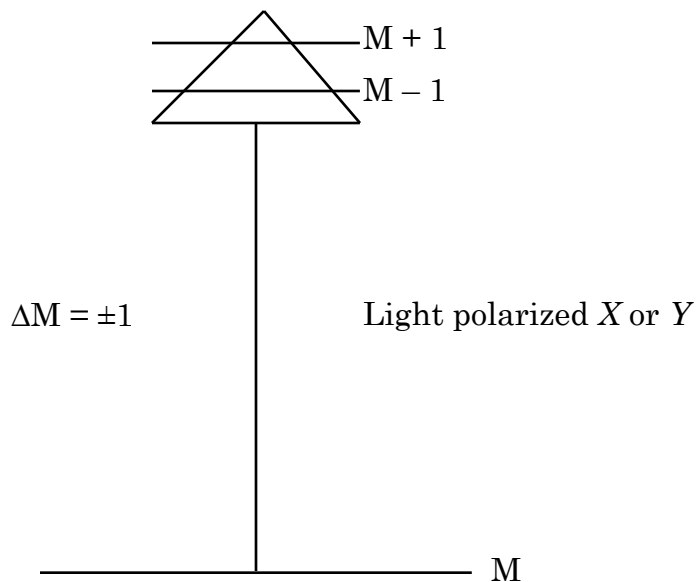
Alternatively, light polarized X or Y (perpendicular to B_Z)

get $M' - M'' = \pm 1$

If you know the upper and lower state patterns, you can disentangle the spectrum and get $g_{J'}$ and $g_{J''}$.

Use a pulsed laser

Excite



Get quantum beats \rightarrow determine $g_{J'}$! For many M'' components, get a set of Zeeman Quantum Beats for J' upper levels, but all upper level quantum beats are at the same frequency!

You can show that g_J is the same for e^- and h^+ .

You can use the hole Slater determinant representation to evaluate H^{Zeeman} for more than 1/2 filled shells.

MIT OpenCourseWare
<https://ocw.mit.edu/>

5.73 Quantum Mechanics I
Fall 2018

For information about citing these materials or our Terms of Use, visit: <https://ocw.mit.edu/terms>.